

HOT DIP GALVANISING

WHAT IS IT, WHY GALVANISE & WHAT ALTERNATIVES ARE THERE?

WHAT IS GALVANISING?

Galvanising is the process of applying a layer of zinc on a steel substrate to provide galvanic corrosion protection. Hot Dipped Galvanising is usually the process most readily identified however, there are other known processes by which zinc can be applied to the steel surface.

Galvanising may therefore be achieved in a number of methods, including:

- Hot Dipped Galvanising (HDG) - involves the dipping of prepared steel into a bath of pure molten zinc metal to form an amalgamⁱ of zinc and iron
- Electro-galvanising - is the deposition of a metallic film on a prepared surface by electrolysisⁱⁱ.
- Hot Metal Spray - involves using zinc or zinc alloy wire that is melted in a special spray gun by an electric arc and is then blown onto the prepared surface by compressed air.
- Zinc-Rich Primers - application of anti-corrosive primers containing high levels of powdered zinc metal.

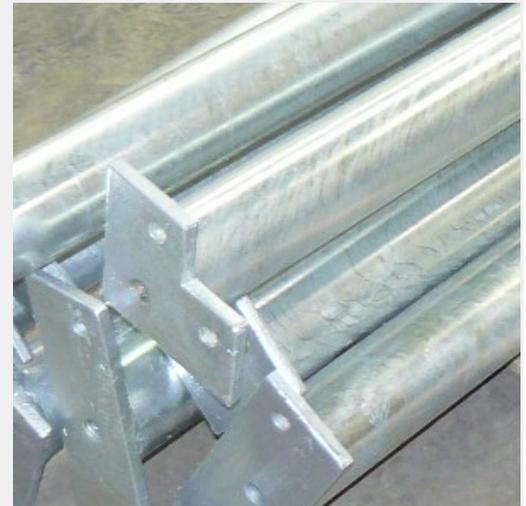
In addition to these processes, there are also different zinc alloy products including Zinalume[®], Duragalv[®], Zincaneal[®], etc.

A micrograph of a cross section of HDG steel shows that an amalgam of steel and zinc exists between the steel substrate and the zinc layer.

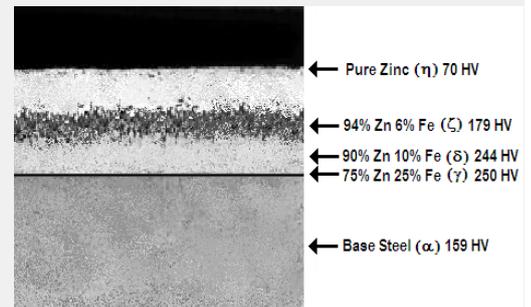
WHY HOT DIP GALVANISE?

The reasons for hot dip galvanising are given below:

- Combats corrosion by the zinc coating galvanicallyⁱⁱⁱ sacrificing itself in place of the mild steel - life span is dependent on environment and thickness of the zinc layer.
- Good for small and difficult to paint items – i.e. steel items that have a high surface area per tonne such as open-web flooring, ladders, handrails, brackets, bolts, girts, purlins etc.
- Provides good anti-corrosive performance in mild/rural environments.
- Provides a hard-wearing surface which is useful in construction. The surface may mar under impact stress but will still serve as an anticorrosive coating.



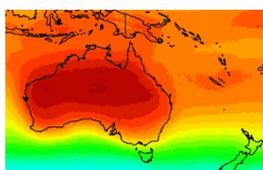
Newly galvanised steel is extremely shiny



Micrograph of hot dip galvanised steel



White rust forms on exposure to the elements



HOT DIP GALVANISING

WHY NOT HOT DIP GALVANISE?

- The HDG zinc layer is sensitive to coastal/marine & industrial (chemical) environments - in these environments the pure zinc surface will sacrifice itself at a relatively **rapid rate**, thus reducing its life considerably.
- Impractical for large or existing structures - requires to be dipped into a bath of molten zinc, thus large items are restricted by the size of the bath and on site application is not possible.
- Hot dip galvanising can **hide defects** and surface contaminants on the surface of the steel.
- The hot zinc bath may **warp** and **distort** items during treatment due to heat involved in the process. This may require extensive reworking. Warping is not always evident until the galvanised sections are delivered to site and the warping prevents assembly!
- A decorative finish may be difficult to achieve, as HDG steel is generally delivered straight to site rather than to a shop, and therefore paint application is done in the open and subject to wind, rain, airborne salts, dust and insects. Spray application on site is difficult to manage, and therefore application is usually limited to brush and roller.
- The HDG steel itself presents **challenges** in surface preparation, defects and contaminants.

WHAT ALTERNATIVES ARE THERE?

As mentioned above, there are three other alternatives. By far the most practical, particularly for steel that is to be painted for decorative purposes, is the application of anti-corrosive primers containing **high levels of zinc metal powder**.

The two most commonly specified zinc-rich primers are:

- inorganic zinc silicate primers, and
- zinc-rich epoxy primers

For more information on the advantages of inorganic zinc silicate coatings and zinc-rich epoxy primers, please refer to Dulux PC Tech Notes 3.8.1 'Inorganic Zinc Silicate Primers' & 3.8.2 'Organic Zinc Rich Primers'.

For information on surface preparation and use of zinc-rich primers on mild steel, please refer to Dulux PC Tech Note 1.1.1 'Mild Steel – What is It?'

For information on challenges concerning the surface preparation of HDG steel, please refer to Dulux PC Tech Note 1.2.2 'Surface Preparation Issues'.

For more information, please contact the Dulux Protective Coatings Technical Consultant in your state.



Hot dip galvanised handrails after four years exposure in a bayside shopping centre



The hot zinc bath may warp and distort items during treatment

ⁱ Amalgam - a mixture or combination. In the case of HDG, the adhesion of the zinc film is a chemical one in that a zinc-iron alloy/mixture is formed immediately on the steel interface.

ⁱⁱ Electrolysis - the decomposition of a chemical compound by an electric current.

ⁱⁱⁱ Galvanic - an electrical current occurring as a result of connecting dissimilar metals. Also known as "Cathodic". Connecting two different metals together (in the presence of moisture and ions) creates an electrical current between the metals, causing one metal to corrode in preference to the other (galvanic corrosion). Contact between mild steel and zinc will cause the zinc to corrode in preference to the mild steel, which effectively protects the steel from corrosion as long as the zinc lasts.